# **PCT**

# WORLD INTELLECTUAL PROPERTY ORGANIZATION International Bureau





# INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Pate C08L 23/16, C		A1 .		I) International Publication Number: WO 98/59002  3) International Publication Date: 30 December 1998 (30.12.98)
(21) International App (22) International Filir			43	(81) Designated States: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GE, GH, GM, GW, HU, ID, IL, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL,
(30) Priority Data: 972714 972726 972727 972729	24 June 1997 (24.06.97) 24 June 1997 (24.06.97) 24 June 1997 (24.06.97) 24 June 1997 (24.06.97)		FI FI FI	TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), Europear patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG).
A/S [DK/DK]; (DK). (72) Inventors; and (75) Inventors/Applica [FI/FI]; Linnu Bo [FI/FI]; H	designated States except US): Be Lyngby Hovedgade 96, DK-280 ants (for US only): PITKANE Therne 15, FIN-07190 Halkia (FI) arjuviita 16 A 27, FIN-02110 E stalo [FI/FI]; Karjapolku 4, F	O Lyng EN, Pa ). MAL spoo (F	ivi M,	Published  With international search report.  Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.
(74) Agents: COCKBA	AIN, Julian et al.; Frank B. Dehn & a Street, London EC4V 4EL (GB).	& Co., 1	179	

- (54) Title: HETEROPHASIC PROPYLENE COPOLYMER AND PROCESS FOR ITS PREPARATION
- (57) Abstract
- A heterophasic polypropylene copolymer having a tensile modulus of 420 MPa or less comprising: i) a semi-crystalline propylene: ethylene: and optionally other  $\alpha$ -olefin polymer matrix; ii) an elastomeric propylene: ethylene and optionally other  $\alpha$ -olefin copolymer.

# FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

	AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
	AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
	AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
	ΑÜ	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
	ΑZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
	BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
	BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
	BE	Belgium	GN	Guinea	MK	The former Yugoslav	TM	Turkmenistan
	BF	Burkina Faso	GR	Greece		Republic of Macedonia	TR	Turkey
	BG	Bulgaria	HU	Hungary	ML	Mali	TT	Trinidad and Tobago
	BJ	Benin	IE	Ireland	MN	Mongolia	UA	Ukraine
	BR	Brazil	IL.	Israel	MR	Mauritania	UG	Uganda
	BY	Belarus	IS	Iceland	MW	Malawi	US	United States of America
	CA	Canada	IT	Italy	MX	Mexico	UZ	Uzbekistan
	CF	Central African Republic	JP	Japan	NE	Niger	VN	Viet Nam
	CG	Congo	KE	Kenya	NL	Netherlands	YU	Yugoslavia
	CH	Switzerland	KG	Kyrgyzstan	NO	Norway	zw	Zimbabwe
	CI	Côte d'Ivoire	KP	Democratic People's	NZ	New Zealand		
	CM	Cameroon		Republic of Korea	PL	Poland		
	CN	China	KR	Republic of Korea	PT	Portugal		
	CU	Cuba	KZ	Kazakstan	RO	Romania		
	CZ	Czech Republic	LC	Saint Lucia	RU	Russian Federation		
	DE	Germany	LI	Liechtenstein	SD	Sudan		
	DK	Denmark	LK	Sri Lanka	SE	Sweden		
	EE	Estonia	LR	Liberia	SG	Singapore		
ĺ						· -		
ı								

#### HETEROPHASIC PROPYLENE COPOLYMER AND PROCESS FOR ITS PREPARATION

The present invention relates to heterophasic polypropylene copolymers and in particular to heterophasic polypropylene copolymers which are soft, have good impact resistance at low temperatures, good heat sealing properties and good optical properties.

Traditionally, polymers in which low stiffness, and therefore low tensile modulus, as well as good impact at low temperatures are desired properties, have been prepared from soft poly(vinyl chloride) (PVC). However, because PVC products cause environmental problems due to emissions of chlorinated organic compounds during combustion there is currently a trend towards replacing PVC with other polymers. As an alternative to PVC polymers polypropylene polymers have been employed since such polymers are known to have suitable resistance to heat and chemicals as well as attractive mechanical properties.

It is known that certain heterophasic polypropylene copolymers are particularly suitable as soft polymers. When producing soft heterophasic polypropylene copolymers, it is usual to prepare a polypropylene copolymer matrix with a comonomer content to obtain the desired properties. To achieve lower stiffness an amorphous elastomeric component can then be added to the copolymer matrix.

For example, in EP-A-0373660 (Himont Incorporated) a propylene polymer composition is described which has good transparency and improved low temperature impact resistance comprising 70 to 98% crystalline copolymer of propylene with ethylene and/or other  $\alpha$ -olefin and an elastomeric propylene-ethylene copolymer.

In EP-A-0416379 (Himont Incorporated) a thermoplastic olefin polymer having elastic properties is disclosed comprising a crystalline polymer fraction

- 2 -

comprising for example a copolymer of propylene with at least one  $\alpha$ -olefin, a semi-crystalline fraction and 2 to 30% amorphous copolymer fraction of  $\alpha$ -olefin and propylene with or without diene.

In the two applications discussed above, for reasons of economy, the initial copolymer matrix can be produced in a slurry reactor rather than in a gas phase reactor. The product of the slurry phase reaction is then flashed to remove unreacted monomers and transferred into a gas phase reactor where further reaction takes place and the elastomeric component is prepared.

However, since the matrix component is prepared in the liquid phase, the comonomer content of the matrix and thus the tensile modulus is limited. Comonomers such as ethylene and other  $\alpha$ -olefins cause swelling of the polymers during the reaction in the slurry reactor. When the reaction medium is flashed to remove the monomer reactants after polymerisation of the matrix but before transfer to the gas phase reactor, the morphology of the particles is destroyed and the bulk density of the powder becomes very low. This sticky material agglomerates on the walls in the flash tank and causes problems in transportation into the gas reactor. These problems increase when the proportion of comonomers in the copolymer increases and therefore the softness of the polymer is limited.

Attempts to reduce the stickiness of the material in the flash tank by catalyst manipulation or by reducing the content of xylene-soluble fraction have had limited success and accordingly, in order to obtain soft polypropylene copolymers with a very high comonomer content gas phase polymerisation has traditionally been required.

It has now been surprisingly found that soft polypropylene copolymers can be prepared economically with low tensile modulus values and high comonomer

content since the flashing of the matrix mixture prepared in a liquid phase reactor (e.g. a slurry reactor) is unnecessary and transfer of the neat reaction mixture from the liquid phase to the gas phase reactor can be effected directly. Since the flashing step is omitted, there are no problems with the sticky material sticking on the walls of the flash tank and therefore higher comonomer concentrations can be achieved and softness properties improved.

Thus, viewed from one aspect the invention provides a heterophasic polypropylene copolymer having a tensile modulus of 420 MPa or less comprising:

- i) a semi-crystalline propylene:ethylene: and optionally other  $\alpha$ -olefin copolymer matrix;
- ii) an elastomeric propylene:ethylene and optionally other  $\alpha$ -olefin copolymer.

Viewed from another aspect the invention provides a process for the preparation of a heterophasic polypropylene copolymer having a tensile modulus of 420 MPa or less comprising:

- i) producing a semi-crystalline propylene: ethylene and optionally other  $\alpha$ -olefin copolymer matrix in one or more slurry reactors and optionally one or more gas phase reactors;
- ii) followed by producing an elastomeric propylene:ethylene and optionally other  $\alpha$ -olefin copolymer in the gas phase;

characterised in that the transfer from liquid phase reactor to a subsequent gas phase reactor is effected without flashing to remove unreacted monomer.

For the purposes of this application, the term copolymer encompasses polymers comprising two or more componers.

The semi-crystalline polypropylene copolymer matrix preferably comprises 0.5 to 10 wt% ethylene and optionally 5 to 12 wt% of other  $\alpha$ -olefin. Where the semi-crystalline polypropylene copolymer matrix

comprises an  $\alpha$ -olefin in addition to ethylene and propylene, ethylene more preferably comprises 1 to 7 wt%, most preferably, 1 to 5 wt% of the matrix and the additional  $\alpha$ -olefin 6 to 10 wt% of the matrix.

Where the semi-crystalline matrix component is an ethylene:propylene copolymer only, the ethylene preferably comprises 3.5 to 8.0 wt%, most preferably 4 to 7 wt% by weight of the matrix.

The other  $\alpha$ -olefin is may be a  $C_{4-20}$  mono or diene, and may be linear, branched or cyclic. The other  $\alpha$ -olefin is preferably of structure  $H_2C$ =CHR where R represents an alkyl group. Preferably, the  $\alpha$ -olefin has between 4 and 8 carbon atoms and is most preferably 1-butene, 1-pentene, 4-methyl-1-pentene, 1-hexene, 1-heptene or 1-octene, especially 1-butene.

The xylene soluble fraction (XS) of the matrix component is preferably in the range 3 to 30%, most preferably 6 to 20% by weight of the matrix component.

The elastomeric propylene:ethylene and optionally other  $\alpha$ -olefin copolymer can comprise any suitable ratio of monomers of propylene, ethylene and optionally other  $\alpha$ -olefin which results in an amorphous or semicrystalline/amorphous elastomeric copolymer. Preferably, the elastomeric component comprises a copolymer of ethylene and propylene only.

The amount of matrix component in the heterophasic copolymers of the invention is between 20 to 90%, preferably 60 to 90% by weight of the heterophasic copolymer. The amount of elastomeric copolymer may be in the range of 10 to 80%, preferably 10 to 40% by weight of the heterophasic polymer. The elastomer component may comprise 95-5 wt%, preferably 95-20 wt% of crystalline phase and 5-95 wt%, preferably 5-80 wt% amorphous phase.

In one embodiment, 5 to 40% wt of elastomer or plastomer, based on the weight of the heterophasic copolymer may be blended into the heterophasic polymers

- 5 -

of the invention. By adding varying amounts of elastomer or plastomer to the heterophasic polymers of the invention, the stiffness of the polymers can be further reduced, optical properties improved and low temperature impact resistance enhanced. Suitable elastomers include ethylene:butene rubber, terpolymer rubber but is preferably ethylene:propylene rubber (EPR). Elastomer may also be added in the form of ethylene:propylene diene monomer (EPDM). elastomers can be prepared by conventional processes and blended into the heterophasic polymers of the invention by standard mixing techniques.

The tensile modulus of the heterophasic polymer of the invention is 420 MPa or less but preferably should be greater than 80 MPa, preferably greater than 100 MPa. More preferably, the tensile modulus should be in the range 100 to 400 MPa, even more preferably 100 to 350 MPa and most preferably 100 to 300 MPa.

The xylene soluble fraction (XS) of the final heterophasic polymer is preferably between 20 to 60%, most preferably 30 to 45%.

To ensure that the heterophasic polymers have suitable impact properties it is preferred that the polymers be classified as ductile at -20°C.

The heterophasic polypropylene copolymers of the invention are produced in a combination of at least one slurry and at least one gas phase reactor connected directly together, thus avoiding the flash step and thus the disadvantages discussed above when producing high comonomer content products using a flash step.

Suitable preparation methods will be readily determined by the person skilled in the art and include but are not limited to:

Α. producing the polypropylene copolymer matrix containing ethylene, and optionally other higher  $\alpha$ -olefin in one or two slurry reactors, then

feeding the slurry reactor mixture directly into a gas phase reactor, and producing in the gas phase reactor (GPR) an elastomeric copolymer.

- B. producing the polypropylene copolymer matrix containing ethylene, and/or other higher α-olefin in two steps firstly in one or more slurry reactors and then in a gas phase reactor using a direct feed from slurry reactor into the GPR, and then feeding the reaction mixture into a second GPR and continuing the polymerisation to produce a similar elastomer as in case A.;
- C. producing a heterophasic polymer as described in A or B and then producing more elastomeric copolymer in further gas phase reactors.
- D. producing a a heterophasic polymer as described in A, B or C, and then blending in a suitable amount of elastomer (e.g. EPR, EPDM) or plastomer prepared by conventional techniques.

The comonomer feeds into the various reactors may be adapted to produce a polymer with the desired properties and the amounts of comonomer will be readily determined by the person skilled in the art.

In the slurry reactor, propylene preferably acts not only as a monomer for polymerisation but also as a diluent. The slurry step, which preferably occurs in a loop reactor, is carried out by feeding into at least one slurry polymerization step a reaction mixture containing 50-95 wt% of propylene, 1-10 wt% of ethylene and 0-40 wt% of other  $\alpha$ -olefin, and a catalyst system whilst maintaining olefin polymerization at a temperature below 75°C. Where two or more slurry reactors are used, these are preferably loop reactors and are fed by the same comonomer mixture as for a

- 7 -

single slurry reactor. From the slurry reactor the reaction mixture is fed directly to one or more gas phase reactors.

Where one gas phase reactor is employed, the gas phase polymerization step is preferably carried out by adding 0-40 wt% for propylene, 1-30 wt% for ethylene, 0-10 wt% for other α-olefin of the feed mixture. Preferably, the gas ratio employed in the feed mixture in the preparation of the elastomeric component is C2/(total monomer) (mol/mol) is between 0.05-0.5, preferably 0.2-0.5 and C4/(total monomer) is greater or equal to 0.15. In this case the gas phase reactor is primarily employed to produce the elastomeric copolymer component which preferably comprises 5 to 40 wt%, most preferably 10 to 40 wt% of the heterophasic polymer of the invention.

Where a further gas phase reactor is employed, the first gas phase reactor is primarily employed to produce further matrix component. In this case the ethylene feed into the first gas phase reactor is preferably 15 wt%, most preferably 1 to 8 wt%. The gas phase polymerization step is preferably continued in the second gas phase reactor by adjusting the gas concentrations to 60-90 wt% for propylene, 5-40 wt% for ethylene and 0-10 wt% of other  $\alpha$ -olefins. Such concentrations produce more elastomeric semicrystalline and amorphous copolymer into the product produced in the first gas phase polymerization step.

In one embodiment, hydrogen may be added into either or both of the slurry phase or gas phase to control the molecular weight of the polymer of the invention. The use of hydrogen in olefin polymerisation is conventional and will be readily applied by the person skilled in the art.

The slurry phase polymerisation may be carried out at temperatures of lower than 75°C, preferably 60-65°C and pressures varying between 30-90 bar, preferably 30-70

- 8 -

bar. The polymerization is preferably carried out in such conditions that 20-90 wt%, preferably 40-80 wt% from the polymer is polymerized in the slurry reactor or reactors. The residence time can be between 15 and 120 min.

The gas phase polymerization step is carried out by transferring the reaction mixture from the slurry phase directly to the gas phase without removing unreacted monomers. , preferably higher than 10 bars. reaction temperature used will generally be in the range 60 to 115°C, preferably 70 to 110°C. The reactor pressure will be higher than 5 bars, and preferably be in the range 10 to 25 bar, and the residence time will generally be 0.1 to 5 hours. Since unreacted monomers from the slurry phase are transferred into the gas phase it is important to establish how much unreacted monomer has been transferred to allow ready determination of how much further monomer to add to the gas phase. measurements can be achieved by simple gas chromatography allowing maintainence of appropriate comonomer concentrations.

The liquid medium from the first stage reactor can function as a cooling medium of the fluid bed in the gas phase reactor, when evaporating therein.

Preferably, a loop reactor is used as said slurry reactor although other reactor types such as a tank reactor could also be employed. According to another embodiment said slurry phase is carried out in two slurry reactors, preferably but not necessarily in two loop reactors. In this way the comonomer distribution can be easily controlled. When continuing the copolymerization in a gas phase reactor or reactors, comonomer content can be increased further. Thus, the matrix polymer can be tailored by adjusting comonomer ratios in different reactors.

Elastomer may be produced into the heterophasic polymer of the invention in one or more gas phase

- 9 -

reactors or blended into the final polymer using standard blending procedures. By controlling the amount and composition of elastomer component in the final polymer the properties, e.g. optical and impact properties and softness can be adjusted.

Polymerisation may be achieved using any standard olefin polymerisation catalyst and these are well known to the person skilled in the art. Preferred catalyst systems comprise an ordinary stereospecific Ziegler-Natta catalyst, metallocene catalysts and other organometallic or coordination catalysts. A particularly preferred catalyst system is a high yield Ziegler-Natta catalyst having a catalyst component, a cocatalyst component, optionally an external donor. The catalyst system may thus contain a titanium compound and an electron-donor compound supported on an activated magnesium dichloride, a trialkylaluminium compound as activator and an electron donor compound.

A further preferred catalyst system is a metallocene catalyst having a bridged structure giving high stereoselectivity and which as an active complex is impregnated on a carrier.

Suitable catalyst systems are described in for example, FI Patent No. 88047, EP 491566, EP 586390 and WO98/12234 which are hereby incorporated by reference.

Soft propylene copolymer products having low stiffness, high low temperature impact and optionally owing good sealability and transparency can be used in a wide variety of applications such as films, moulded items, sheets, lids, bottles, fibres, tubes, foams, cable jacketing and insulation, and compounds (flame retardant and other high filled compounds).

#### **EXAMPLES**

DESCRIPTION OF SOME IN

A combination of a slurry and gas phase reactors were used for producing soft heterophasic polypropylene copolymers of the invention.

The following analytical techniques were used in testing the polymers synthesised:

Melt flow rates (MFR) were measured under a load of 2.16 kg and at 230°C according to ISO 1133.

Comonomer contents (ethylene and butene) were measured with Fourier transform infrared specroscopy (FTIR) calibrated by nuclear magnetic resonance spectroscopy (NMR).

Melting temperatures (peak temperatures) were measured with differential scanning calorimetry (DSC) using a temperature increasing rate of 10°C/min.

Tensile strength (tensile stress at yield) was measured according to ISO 572-2 (cross head speed = 50 mm/min).

Tensile modulus was measured according to ISO 572-2 (cross head speed = 1 mm/min).

Flexural modulus was measured according to ISO 178.

Izod, notched impact strength was measured according to ISO 180/1A.

Instrumentated falling weight impact was measured according to ISO 6603-2.

Film tests were measured from 40 micrometer thin film produced at a cast film line (Collin), melt temperature 270°C and chill roll temperature 30°C.

Film modulus (1% secant modulus) was measured according to ISO 1184 (ASTM D882) (cross head speed = 5 mm/min).

Haze was measured according to ASTM D 1003 and gloss at 60°C according to ASTM D 523.

Heat sealing properties were measured from 25 micrometer ABA-film (core layer PP homopolymer and as the surface layer the examined heat seal PP terpolymer. Sealing time was 0.1 s and pressure 3 bar. Heat seal strength = pealing of the seal using film strips, broadness 25.41 mm was measured at an Instron universal tester (cross head speed = 100 mm/min).

#### Examples 1- 5:

Polymerisations were carried out in a stirred 5 dm<sup>3</sup> bench scale reactor. A matrix was produced in liquid propylene at 60°C in the first stage. After the desired 1h polymerisation time the pressure was reduced slowly to 7 bar and polymerisation was continued in gas phase. In gas phase, semicrystalline and amorphous copolymers were polymerised at 75°C into the matrix 1 to 1.5h.

The catalyst used was prepared according to WO98/12234, the activity of which was 35 kg PP/g cat h. The polymerization conditions and the polymer

properties are presented in Table 1.

In the Example 5, the pressure was reduced rapidly to atmospheric pressure, simulating flash tank operation. Polymerisation in the gas phase was not possible and only the matrix component was produced.

Table 1

	EX 1	EX 2	EX 3	EX 4	EX 5
Slurry Stage				<del>                                     </del>	
C <sub>2</sub> feed, w%	3.2	3.2	2.4	1.5	2.6
C, feed, w%	-	-	20	25	20
Gas Phase Stage					
C <sub>2</sub> /total monomer	0.33	0.34	0.31	0.36	-
in reactor					
C4 feed	-	-	C4 from	C4 from	-
			1. stage	1. stage	
Matrix					
C2, W8	4.8	4.8	4.0	2.8	2.8
C1, W8	-	-	6.4	8.3	6.8
Xylene Soluble	10.0	10.02	20.6	19.7	15.8
Fraction (XS), %			[   _		·
Final Product					
Amt of matrix, %	75	70	73	79	100
MFR, g/10 min	4.2	3.3	2.6	5.3	4.5
C <sub>2</sub> , w%	14.1	15.9	12.1	11.8	
C., w%		_	7.3	9.4	
XS, %	32.1	35.7	40.2	42	
IV/XS, dl/g	2.0	2.0	1.7	1.8	
C <sub>2</sub> /XS, %	30.4	27.6	20.3	n.a.	
C <sub>4</sub> /XS, %	-	-	12.5	n.a.	
Tensile modulus,	340	260	260	200	
MPa					
falling weight	ductile	ductile	ductile	ductile	
impact at -20°C					

n.a. = not analysed.

# Example 6 (comparitive)

A combination of a pilot-scale loop reactor and a gas phase reactor were used to produce soft heterophasic polypropylene copolymers. Propylene, ethylene and hydrogen were fed into the loop reactor. The polymerization temperature in the loop reactor was 70°C. The temperature in the GPR was 70°C. The pressure in the loop reactor was 40 bar and in the gas phase reactor 10 bar. The catalyst used was as for Examples 1 to 5 giving an activity of 35 kg PP/g cat h.

The random copolymer produced in the loop reactor gave a ethylene content of 3.5-3.7 wt% and a xylene soluble fraction (XS) = 9 wt%.

When this copolymer was flashed severe problems arised by sticking in the flash tank, which made the transport to the GPR impossible.

#### Example 7

A combination of a pilot scale loop reactor and a gas phase reactor was used to produce soft heterophasic copolymer. Random ethylene:propylene copolymer was produced in the loop reactor. The ethylene content of the polymer was 4.2 wt% and the XS was 9 wt%. This reactor mixture was transported directly into the gas phase reactor. More ethylene and propylene were fed using gas ratio C2/(total monomer) of 0.35 mol/mol into the GPR and impact copolymer was produced.

The final heterophasic polymer had ethylene content of 9.5% and XS-fraction of 25%.

To further improve the softness and impact properties 15% of very low density PE copolymer (VLDPE) (Exact 2M011 from DEX plastomers, density 0.88) was blended giving the following properties:

Table		+VLDPE (15%)
Tensile modulus, MPa Falling weight impact	590 ductile	420 ductile
at -20 C, failure type falling weight impact	brittle/ductile-	ductile
at -40 C, failure type	brittle	

- 14 -

Film modulus, MPa 280 210

#### Example 8

A high ethylene content propylene random copolymer was produced in pilot reactors; a lower ethylene content random copolymer was produced in the loop. The reaction mixture was transported into the gas phase reactor and more ethylene was fed in.

To further improve the softness and impact properties 20% of very low density PE copolymer (VLDPE) (Eact 2M011 from DEX plastomers, density 0.88) was blended giving the following properties:

Property		+ VLDPE (20 %)
C2 in first stage, wt%	3.8	
XS of first stage, wt%	6.7	
total ethylene content, wt%	7.9	26.5
Tensile modulus, MPa	380	280
falling weight impact,	ductile	ductile
at 0 C, failure type falling weight impact,	brittle	ductile
	DITCCIC	ductile
at -20 C, failure type		
film modulus, MPa	220	190
haze of film, %	0.5	2.2
gloss of film	150	140

## Example 9

A heterophasic polypropylene was produced in the pilot scale reactors; a random copolymer was produced in the loop, this reactor mixture was transported directly into the gas phase reactor where more ethylene and propylene were fed using a gas ratio, C2/total monomer of 0.35.

First stage:

C2, wt% 4.2 XS, wt% 7.4

Final product:

C2, wt% 19.3 XS, wt% 40 Tensile modulus, MPa 360 Falling weight impact, at -20, failure type ductile Falling weight impact, at -40, failure type ductile Izod impact, at -30, kJ/m2 22 Film modulus, Mpa 190

#### Example 10

Haze, 40 um film

A continuously operated pilot plant was used to produce PP-copolymer with good impact and creep properties. The plant comprises catalyst, alkyl, donor, propylene and ethylene feed systems, a prepolymerisation reactor, a loop reactor and two fluidized bed gas phase reactors (GPR).

7.6

The catalyst, alkyl, donor and propylene are fed to the prepolymerisation reactor. The polymer slurry from the prepolymerisation reactor is fed to a loop reactor to which also hydrogen, ethylene and additional propylene are fed.

The polymer slurry fiom the loop reactor and additional hydrogen and propylene are fed to the first GPR. The polymer from the first GPR is fed to the second GPR. Ethylene, some hydrogen and additional propylene was fed to the second GPR. The polymer formed and unreacted propylene are separated after removal from the second GPR.

The catalyst used is a highly active and stereospecific

- 16 -

ZN-catalyst made according to U.S. Patent No. 5,234,879. The catalyst is contacted with triethylaluminium (TEA) and dicyclopentyldimethoxysilane (DCPDMS) (Al/Ti ratio is 150 and Al/Do 10 (mole)) before feeding to the prepolymerisation reactor.

The catalyst is fed according to U.S. Patent No. 5,385,992 and is flushed with propylene to the loop reactor. The prepolymerirnfion reactor is operated at a pressure of 51 bar, a temperature of 20°C and a mean residence time of the catalyst of 7 min.

The loop reactor is operated at a 50 bar pressure, a 75°C temperature and a mean residence time of the catalyst of 1 h. The MFR (2.16 kg, 230 °C) of the PP-random polymer produced in the loop is adjusted to 7 by controlling the hydrogen feed. The ethylene content is adjusted to 3.5% w/w by using the ethylene feed as a control means.

The polymer slurry from the loop reactor is transferred to the first GPR. The first GPR reactor is operated at a total pressure of 29 bar and a partial propylene pressure of 21 bar. The operational temperature is 80°C and the mean residence time of the catalyst 1.5 h. The MFR (2.16 kg, 230°C) of the PP-randompolymer taken out of the GPR is adjusted to 10 by using the partial hydrogen pressure. The ethylene content is set at 2% w/w by adjusting the production split between the reactors.

The polymer from the first GPR is transferred to the second GPR. The second GPR is operated at a total pressure of 10 bar and a partial monomer pressure of 7 bar. The operational temperature is 80°C and mean residence time of the catalyst 1.5h. The MFR (2.16 kg, 230°C) of the PP-copolymer taken out of the GPR is adjusted to 7 via the partial pressure of the hydrogen. The ethylene content is set at 10% w/w by adjusting the

- 17 -

partial pressure of ethylene and by controlling the 'production split between the reactors.

The desired properties are achieved with a production split of 1% in prepolymerization, 40% in loop, 40% in the first GPR and 19% in the second GPR.

#### Example 11

A continuously operated pilot plant was used to produce very soft PP-copolymer. The plant comprises catalyst, alkyl, donor, propylene and ethylene feed systems, prepolymerisation reactor, loop reactor and a fluidized bed gas phase reactor (GPR).

The catalyst, alkyl, donor and propylene were fed to the prepolymerisation reactor. The polymer slurry from the prepolymerisation reactor was fed to a loop reactor to which also hydrogen, ethylene and additional propylene was fed.

The polymer slurry from the loop reactor and additional ethylene, hydrogen and propylene were fed to the GPR.

The formed polymer and unreacted monomers were separated after removal from GPR.

The catalyst used was a highly active and stereospecific ZN-catalyst made according to U.S. Patent No. 5,234,879. The catalyst was contacted with triethylaluminium (TEA) and dicyclopentyldimethoxysilane (DCPDMS) (AI/Ti ratio was 150 and Al/Do 10 (mole)) before feeding to the prepolymerisation reactor.

The catalyst was fed according to U.S. Patent 5,385,992 and was flushed with propylene to the loop reactor. The prepolymerisation reactor was operated at a 51 bar pressure, a 20°C temperature and a mean residence time of

- 18 -

the catalyst of 7 min.

The loop reactor was operated at a pressure of 50 bar, a temperature of 75°C temperature and mean residence time of the catalyst at 1 h. The MFR (2.16 kg, 230 °C) of the PP-randompolymer produced in the loop was controlled to be 4 via hydrogen feed. The ethylene content was adjusted to 3.8% w/w by controlling the ethylene feed.

The polymer slurry from the loop reactor was transferred to the first GPR. The first GPR reactor was operated at a total pressure of 29 bar and a partial propylene pressure of 21 bar. The operational temperature was 80°C and the mean residence time of the catalyst 1.2 h. The MFR (2.16 kg, 230°C) of the PP-randompolymer taken out of the GPR was set at 2.5 by adjusting the partial hydrogen pressure. The ethylene content was set at 8% w/w by adjusting production split between the reactors and the partial pressure of ethylene.

Desired properties are achieved with a production split of 1% in prepolymerisation, 45% in loop and 55% in the GPR.

#### Example 12

A pilot plant operated continuously is used to produce PP-copolymer with good impact and creep properties. The plant comprises a catalyst alkyl, donor, propylene and ethylene feed systems, a prepolymerization reactor, a loop reactor and two fluidized bed gas phase reactors (GPR).

The catalyst, alkyl, donor and propylene are fed to the prepolymerization reactor. The polymer slurry from the prepolymerization reactor is fed to a loop reactor in which also hydrogen, ethylene and additional propylene

- 19 -

are fed.

The polymer slurry from the loop reactor and additional hydrogen and propylene are fed to the 1st GPR. The polymer from 1st GPR is fed to 2nd GPR. Ethylene, some hydrogen and additional propylene was fed to the second GPR. The formed polymer and unreacted propylene are separated after removal from 2nd GPR.

The catalyst used is a highly active and stereospecific ZN-catalyst made according to U.S. Patent 5,234,879. The catalyst is contacted with triethylaluminium (TEA) and dicyclopentyldimethoxysilane (DCPDMS) (Al/Ti ratio is 150 and Al/Do 10 (mole)) before feeding to the prepolymerization reactor.

The catalyst is fed according to U.S. Patent 5,385,992 and is flushed with propylene to the loop reactor. The prepolymerization reactor is operated at a pressure of 51 bar, a temperature of 20 °C and a mean residence time of the catalyst of 7 min.

The loop reactor is operated at 50 bar pressure, 75 °C temperature and mean residence time of the catalyst at 1 h. The MFR (2.16 kg, 230 °C) of the PP-randompolymer produced in the loop is controlled to be 7 via hydrogen feed. The ethylene content is controlled to be 3.5 % w/w via ethylene feed.

The polymer slurry from the loop reactor is transferred to the 1st GPR. The 1st GPR reactor is operated at 29 bar total pressure and 21 bar partial pressure of propylene, 80 °C temperature and mean residence time of the catalyst at 1.5 h. The MFR (2.16 kg, 230 °C) of the PP-randompolymer taken out of the GPR is controlled to be 10 via partial pressure of hydrogen. The ethylene content is controlled to be 2% w/w via production split

- 20 -

between the reactors.

The polymer from the 1st GPR is transferred to the 2nd GPR. The 2nd GPR reactor is operated at 10 bar total pressure and 7 bar partial pressure of monomers, 80°C temperature and mean residence time of the catalyst of 1.5 h. The MFR (2.16 kg, 230 °C) of the PP-copolymer taken out of the GPR is controlled to be 7 via partial pressure of hydrogen. The ethylene content is controlled to be 10 % w/w via partial pressure of ethylene and controlling the production split between the reactors.

Desired properties are achieved with production split 1% in prepolymerisation, 40% in loop and 40% in the first GPR and 19% in the second GPR.

#### Example 13

A pilot plant operated continuously is used to produce PP-copolymer with good impact and creep properties. The plant comprises a catalyst, alkyl, donor, propylene and ethylene feed systems, a prepolymerisation reactor, a loop reactor and 2 fluidized bed gas phase reactors (GPR).

Catalyst, alkyl, donor and propylene are fed to the prepolymerisation reactor. The polymer slurry from the prepolymerisation reactor is fed to a loop reactor into which also hydrogen, ethylene and additional propylene are fed.

The polymer slurry from the loop reactor and additional hydrogen and propylene are fed to the 1st GPR. The polymer from the 1st GPR is fed to 2nd GPR. Ethylene, some hydrogen and additional propylene was fed to the second GPR. The formed polymer and unreacted propylene are separated after removal from the 2nd GPR.

The catalyst used is a highly active and stereospecific ZN-catalyst made according to U.S. Patent No. 5,234,879. The catalyst is contacted with triethylalunnium (TEA) and dicyclopentyldimethoxysilane (DCPDMS) (Al/Ti ratio is 150 and Al/Do 10 (mole)) before feeding to the prepolymerisation reactor.

The catalyst is fed according to U.S. Patent No. 5,385,992 and is flushed with propylene to the loop reactor. The prepolymerrrafion reactor is operated at a pressure of 51 bar and a temperature of 20°C and a mean residence time of the catalyst amounting to 7 min.

The loop reactor is operated at a 50 bar pressure, 75°C temperature and mean residence time of the catalyst at 1 h. The MFR (2.16 kg, 230 °C) of the PPrandompolymer produced in the loop is controlled to be 7 via hydrogen feed. The ethylene content is controlled to be 3.5 % w/w via ethylene feed.

The polymer slurry from the loop reactor is transferred to the 1st GPR. The 1st GPR reactor is operated at 29 bar total pressure and 21 bar partial pressure of propylene, 80 °C temperature and mean residence time of the catalyst at 1.5 h. The MFR (2.16 kg, 230 °C) of the PP-randompolymer taken out of the GPR is controlled to be 10 via partial pressure of hydrogen. The ethylene content is controlled to be 2% w/w via production split between the reactors.

The polymer from the 1st GPR is transferred to the 2nd GPR. The 2nd GPR reactor is operated at a total pressure of 10 bar and a partial monomer pressure of 7 bar, a temperature of 80 °C and a mean residence time of the catalyst of 1.5 h. The MFR (2.16 kg, 230 °C) of the PP-copolymer taken out of the GPR is adjusted to 7 using partial pressure of hydrogen. The ethylene content is

adjusted to 10 % w/w via partial pressure of ethylene and controlling the production split between the reactors.

The desired properties are achieved with a production split comprising 1% in prepolymerrzation, 40% in loop and 40% in the first GPR and 19% in the second GPR.

### Example 14

A pilot plant operated continuously was used to produce very soft PP-copolymer. The plant comprises a catalyst, alkyl, donor, propylene and ethylene feed systems, prepolymerisation reactor, loop reactor and a fluidized bed gas phase reactor (GPR).

Catalyst, alkyl, donor and propylene was fed to the prepolymerisation reactor. The polymer slurry from the prepolymerisation reactor was fed to a loop reactor in which also hydrogen, ethylene and additional propylene was fed.

The polymer slurry from the loop reactor and additional ethylene, hydrogen and propylene were fed to the GPR. The formed polymer and unreacted monomers were separated after removal from GPR.

The catalyst used was a highly active and stereospecific ZN-catalyst made according to U.S. Patent 5,234,879. The catalyst was contacted with triethylalunnnium (TEA) and dicyclopentyldimethoxysilane (DCPDMS) (Al/Ti ratio was 150 and Al/Do 10 (mole)) before feeding to the prepolymerisation reactor.

The catalyst was fed according to U.S. Patent 5,385,992 and was flushed with propylene to the loop reactor. The prepolymerisation reactor was operated at 51 bar pressure, 20°C temperature and mean residence time of the

catalyst at 7 min.

The loop reactor was operated at 50 bar pressure,  $75^{\circ}$ C temperature and mean residence time of the catalyst at 1 h. The MFR (2.16 kg, 230  $^{\circ}$ C) of the PPrandompolymer produced in the loop was controlled to be 4 via hydrogen feed. The ethylene content was controlled to be 3.8 % w/w via ethylene feed.

The polymer slurry from the loop reactor was transferred to the 1st GPR. The 1st GPR reactor was operated at 29 bar total pressure and 21 bar partial pressure of propylene, 80 °C temperature and mean residence time of the catalyst at 1.2 h. The MFR (2.16 kg, 230 °C) of the PPrandompolymer taken out of the GPR was controlled to be 2.5 via partial pressure of hydrogen. The ethylene content was controlled to be 8% w/w via production split between the reactors and partial pressure of ethylene.

Desired properties are achieved with a production split of 1% in prepolymerisation, 45% in loop and 55% in the GPR.

The polymer from the GPR could have been transferred to another GPR to produce even softer PP copolymer by having even higher partial pressure of ethylene in the 2nd GPR.

#### Examples 15-18

A combination of a pilot-scale loop reactor and a gas phase reactor were used to produce propylene copolymers for films. Propylene, ethylene, butene and hydrogen were fed into the loop reactor. The polymerisation temperature was 60°C in both reactors. The pressure in the loop reactor was 35 bar and in the gas phase reactor 15 bar. The catalyst used was a prepolymerised catalyst

prepared according to F170028 and FI 86472, the activity of which was 35 kg PP/g cat h.

The polymer produced along with unreacted monomers was flashed straight into the gas phase reactor and polymerisation was completed therein. The polymerisation conditions are presented in the following table.

	Ε×	15	Ê×	16	Ex	17'	£×	18
	loop	gas	loop	gas	loop	gas	юор	gas
Pressure bar	35	15	35	15	35	15	35	35
Temperature °C	60	60	60	60	60	60	60	60
Production rate	8.0	1.0	8.0	5	8.0	5	6.7	5.0
Propyle feed kg/h	24	-	24	-	23		22	·
Ethylene feed kg/h	0.27	0.15	0.27	0.2	0.25	0.2	0.21	0.17
Butene feed kg/h	5.6	-	5.6	-	7.3	-	8.0	-
Residence time min	150	10	150	120	150	120	150	150
Bulk density g/dm <sup>3</sup>		0.42		0.39		0.4		0.4

The product properties of the copolymer obtained from the gas phase reactor are presented the following Table.

	Ex 15	Ex 16	Ex 17	Ex 18
MFR (g/10 min)	6.2	5.5	4.9	5.6
Total ethylene content (w-%)	2.7	4.3	3.2	2.9
Total butene content (w-%)	5.9	8.0	10.3	11.7
Melting temperature (°C)	131	128	125	124
Izod notched at RT (kJ/m²)	9.7	34.5	27	12.5
Tensile strength (MPa)	21.0	15.5	15.9	16.0
Film tests (thickness 40 µm)				
1 % secant modulus (MPa)	310	200	220	250
Haze (%)	1.1	1.3	1.1	1.1
Gloss, 60 °	148	145	149	147
Heat seal strength: Temp./seal strength		120°C/4.9 N		110 °C/4,7 N

# Example 19

Polymerisation was carried out as in Examples 15 to 18 by using a wax-prepolymerized catalyst made according to Finnish patent 88047. The polymerisation conditions are presented in the following Table.

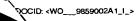
	E×	19
	loop	gas
Temperature (*C)	60	74
Pressure (bar)	38	15
Production rate (kg/h)	11.0	8
Propylene feed (kg/h)	23	-
Ethylene (ccd (kg/h)	0.25	-
Butene feed (kg/h)	7.3	
Residence time min	114	120
Bulk density (kg/dm <sup>3</sup> )		0.35

The copolymer obtained had the properties shown in the following Table.

	Ex 19
MFR <sub>2</sub> (g/10 min)	4.9
Ethylene content (w-% w-w)	3.5
Butene content (w-w-%)	12.4
Melt temperature (°C)	120
Tensile stregth (MPa)	14.2
Tensile modulus (MPa)	360
Izod notched kJ/m²	54
1 % secarst modulus	190
Haze (%)	1.3
Gioss, 60	152
Heat seal strength temp./seal strength	

Claims

- 1. A heterophasic polypropylene copolymer having a tensile modulus of 420 MPa or less comprising:
- i) a semi-crystalline propylene:ethylene: and optionally other  $\alpha$ -olefin polymer matrix;
- ii) an elastomeric propylene: ethylene and optionally other  $\alpha$ -olefin copolymer.
- 2. A polymer as claimed in claim 1 wherein said matrix comprises 0.5 to 10 wt% ethylene and 5 to 12% wt additional  $\alpha$ -olefin.
- 3. A polymer as claimed in claim 2 wherein said matrix comprises 1 to 7 wt% ethylene and 6 to 10% wt additional  $\alpha$ -olefin.
- 4. A polymer as claimed in anyone of claims 1 to 3 wherein said matrix comprises a further  $\alpha$ -olefin having 4 to 8 carbons.
- 5. A polymer as claimed in of claim 4 wherein said matrix comprises 1-butene, 1-pentene, 4-methyl-1-pentene, 1-hexene, 1-heptene or 1-octene.
- 6. A polymer as claimed in of claim 5 wherein said matrix comprises 1-butene.
- 7. A polymer as claimed in claim 1 wherein said matrix component is an ethylene:propylene copolymer only.
- 8. A polymer as claimed in claim 7 wherein said matrix component comprises 3.5 to 8 wt% ethylene.
- 9. A polymer as claimed in claim 8 wherein said matrix component comprises 4 to 7 wt% ethylene.

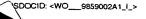


- 10. A polymer as claimed in any one of claims 1 to 9 wherein the xylene soluble fraction (XS) of the matrix component is in the range 3 to 30 wt%.
- 11. A polymer as claimed in claim 10 wherein the xylene soluble fraction (XS) of the matrix component is in the range 6 to 20 wt%.
- 12. A polymer as claimed in any one of claims 1 to 11 wherein said matrix component forms 20 to 90 wt% of the heterophasic polymer.
- 13. A polymer as claimed in claim 12 wherein said matrix component forms 60 to 90 wt% of the heterophasic polymer.
- 14. A polymer as claimed in any one of claims 1 to 13 wherein said elastomeric component comprises a further alpha olefin.
- 15. A polymer as claimed in any one of claims 1 to 13 wherein said elastomeric component is a propylene:ethylene copolymer only.
- 16. A polymer as claimed in any one of claims 1 to 15 wherein said elastomeric component forms 10 to 80 wt% of the heterophasic polymer.
- 17. A polymer as claimed in claim 16 wherein said elastomeric component forms 10 to 40 wt% of the heterophasic polymer.
- 18. A polymer as claimed in any one of claims 1 to 17 wherein the tensile modulus of the polymer is in the range 100 to 400 MPa.
- 19. A polymer as claimed in claim 18 wherein the

tensile modulus of the polymer is in the range 100 to 300 MPa.

- 20. A polymer as claimed in any one of claims 1 to 11 wherein the xylene soluble fraction (XS) of the heterophasic polymer is between 20 to 60%.
- 21. A polymer as claimed in claim 20 wherein the xylene soluble fraction (XS) of the heterophasic polymer is between 30 to 45%.
- 22. A polymer as claimed in any one of claims 1 to 12 wherein 5 to 40 wt% of further elastomer, based on the weight of the heterophasic copolymer, is blended into the polymer.
- 23. A polymer as claimed in claim 22 wherein said further elastomer is ethylene:propylene rubber or plastomer.
- 24. A process for the preparation of a heterophasic polypropylene copolymer as claimed in anyone of claims 1 to 14 comprising the steps of:
- i) producing a semi-crystalline propylene: ethylene and optionally other  $\alpha$ -olefin copolymer matrix in one or more slurry reactors and optionally one or more gas phase reactors;
- ii) followed by producing an elastomeric propylene:ethylene and optionally other  $\alpha$ -olefin copolymer in the gas phase;

characterised in that the transfer from liquid phase reactor to a subsequent gas phase reactor is effected without flashing to remove unreacted monomer.



25. Use of a polymer as claimed in anyone of claims 1 to 23 in the manufacture of films, moulded items, sheets, lids, bottles, fibres, tubes, foams, cable jacketing, insulation, flame retardant and other high filled compounds.

Jonal Application No

PCT/GB 98/01843 . CLASSIFICATION OF SUBJECT MATTER PC 6 C08L23/16 C08F IPC 6 C08F297/08 According to International Patent Classification (IPC) or to both national classification and IPC **B. FIELDS SEARCHED** Minimum documentation searched (classification system followed by classification symbols) IPC 6 C08F C08L Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practical, search terms used) C. DOCUMENTS CONSIDERED TO BE RELEVANT Citation of document, with indication, where appropriate, of the relevant passages Category ' Relevant to claim No. X EP 0 603 723 A (HIMONT INC) 29 June 1994 . 1,4-7,12-1416,17,25 see page 2, line 7 - line 11 see page 2, line 47 - page 3, line 13; claims X EP 0 373 660 A (HIMONT INC) 20 June 1990 1,4-6, cited in the application 12-14, 16, 17, 25 see page 2, line 46 - page 3, line 2 see page 3, line 12 - line 25 see page 2, line 8 - line 11; claims X EP 0 416 379 A (HIMONT INC) 13 March 1991 1,4-7, cited in the application 14,25 see page 4, line 10 - line 12 see page 3, line 12 - line 42 Further documents are listed in the continuation of box C. Patent family members are listed in annex. Special categories of cited documents: "T" later document published after the international filing date or priority date and not in conflict with the application but "A" document defining the general state of the art which is not considered to be of particular relevance cited to understand the principle or theory underlying the earlier document but published on or after the international "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to filing date 1. document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another involve an inventive step when the document is taken alone citation or other special reason (as specified) "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such docu-"O" document referring to an oral disclosure, use, exhibition or other means ments, such combination being obvious to a person skilled document published prior to the international filing date but later than the priority date claimed "&" document member of the same patent family Date of the actual completion of theinternational search Date of mailing of the international search report 14 October 1998 28/10/1998 Name and mailing address of the ISA Authorized officer European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nt, Kaumann, E Fax: (+31-70) 340-3016

Form PCT/ISA/210 (second sheet) (July 1992)

Inte jonal Application No PCT/GB 98/01843

C /Continue	Mines DOCUMENTO CONCUERTO TO DE DE CONCUE	PC1/GB 98/01843
Category *	ation) DOCUMENTS CONSIDERED TO BE RELEVANT  Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0 714 923 A (SHOWA DENKO KK)	1,25
	5 June 1996 see page 20, line 56 — line 58; claims ———	
X	DE 44 26 569 A (UBE INDUSTRIES) 2 February 1995 see column 10, line 20 - line 34; claims	1
X	US 4 740 550 A (FOSTER E GORDON) 26 April 1988 cited in the application see the whole document see column 2, line 37 - line 39 see column 3, line 31 - line 35	24
X	US 4 740 551 A (FOSTER E GORDON) 26 April 1988 see column 8, line 7 - line 8 see the whole document	24
X	WO 88 02376 A (UNION CARBIDE CORP) 7 April 1988 cited in the application see page 7, line 30 - page 11, line 21 see claims; examples	24

Information on patent family members

Inte \_ional Application No PCT/GB 98/01843

Detect description		1017 GB	96/01643	
Patent document cited in search report	Publication date	Patent family member(s)	Publication date	
EP 0603723 A	29-06-1994	IT 1256663 B AT 140947 T AU 667410 B AU 5236493 A DE 69303881 D DE 69303881 T ES 2090829 T JP 7003087 A MX 9307889 A US 5541260 A	12-12-1995 15-08-1996 21-03-1996 30-06-1994 05-09-1996 16-01-1997 16-10-1996 06-01-1995 29-07-1994 30-07-1996	
EP 0373660 A	20-06-1990	AT 133971 T AU 616330 B AU 4676389 A CA 2005537 A CN 1045983 A CZ 8907079 A DE 68925628 D DE 68925628 T ES 2085270 T FI 101074 B HU 212449 B IL 92654 A JP 2258854 A MX 18699 A PT 92582 A, B RU 2043373 C US 5298561 A	15-02-1996 24-10-1991 21-06-1990 14-06-1990 10-10-1990 11-06-1997 21-03-1996 05-09-1996 01-06-1996 15-04-1998 28-06-1996 12-04-1994 19-10-1990 31-01-1994 29-06-1990 10-09-1995 29-03-1994	
EP 0416379 A	13-03-1991	AT 136911 T AU 630731 B AU 6120890 A CA 1331239 A CN 1051183 A DE 69026558 D DE 69026558 T DK 416379 T HU 210337 B JP 3097747 A NO 177789 B	15-05-1996 05-11-1992 28-02-1991 02-08-1994 08-05-1991 23-05-1996 02-10-1996 13-05-1996 28-03-1995 23-04-1991 14-08-1995	

information on patent family members

Inte Jonal Application No PCT/GB 98/01843

Patent document cited in search report	t	Publication date		atent family nember(s)	Publication date	
EP 0416379	A		PT RU US	95063 A 2083611 C 5326639 A	18-04-1991 10-07-1997 05-07-1994	
EP 0714923	Α	05-06-1996	AU CA CN JP US US	3790395 A 2162946 A 1128767 A 8198922 A 5703172 A 5804665 A 8198923 A	30-05-1996 22-05-1996 14-08-1996 06-08-1996 30-12-1997 08-09-1998 06-08-1996	
DE 4426569	Α	02-02-1995	JP US	7090035 A 5468810 A	04-04-1995 21-11-1995	
US 4740550	Α	26-04-1988	NONE			
US 4740551	Α	26-04-1988	NONE			
WO 8802376	A	07-04-1988	AT AU CA CN CZ DE DE EP FI JP	137507 T 604271 B 8104287 A 1284549 A 1023806 B 8706960 A 3751792 D 3751792 T 0283512 A 882462 A,B,	15-05-1996 13-12-1990 21-04-1988 28-05-1991 16-02-1994 15-02-1995 05-06-1996 02-10-1996 28-09-1988 25-05-1988	